DR. BABASAHEB AMBEDKAR TECHNOLOGICAL UNIVERSITY, LONERE								
	Regular/Supplementary Winter Examination – 2024							
Course	Course: B.Tech. Branch: Mechanical Engineering/Mechanical Engineering (Sandwich)							
Subject Code & Name: BTMES304; Materials Science and Metallurgy Semester : III								
Max Marks: 60 Date: 12/02/2025					Duration: 3 Hr.			
 Instructions to the Students: 1 Each question carries 12 marks. 2. Question No. 1 will be compulsory and include objective-type questions. 3. Candidates are required to attempt any four guestions from Question No. 2 to Question No. 6. 4 The level of question/expected answer as per OBE or the Course Outcome (CO) on which the question is based is mentioned in () in front of the question. 5. Use of non-programmable scientific calculators is allowed. 6. Assume suitable data wherever necessary and mention it clearly. 								
					(Level/CO)	Marks		
Q. 1	Objective type q	uestions. (Compul	sory Question)			12		
1	There are fourtee	en atoms in a unit o	cell of		C01	1		
14	a. BCC	b. FCC	c. HCP	d. none of these		14		
20	The elastic stress	strain behavior of	rubber is		C01			
38	a linear	h Non-linear	c Plastic	d No fixed	-	32		
10	a. Lineai	b. Non-intear		relationshin		01		
LC 3	Substitution of a	foreign atom in th	e site of parent at	m in the crystal	C01	Ω 1		
5	is a					-		
	a. Vacancy	b. Substitution	c. Volume	d. Line				
	defect	impurity	imperfection	imperfections				
4	Gibbs phase rule	curs	CO3	1				
4	a. F – C + P = 2	b. F + C – P = 2	c. F + C + P = 2	d. F – C – P = 2	,	4		
5	The percentage of	of carbon in cast iro	on varies from	I	CO2			
30	a. 0.1 to 0.5	b. 0.5 to 1	c. 0.5 to 1	d. 2 to 4.5		202		
6	6 Steel containing 0.8 to 1.5% carbon, is known as					0 1		
2 L	a. mild steel	b. dead mild	c. medium	d. high carbon		S		
		steel	carbon steel	steel				
7	7 A given component cracked after heat treatment. What can be the					1		
	possible reason?							
	a. It was heated	b. It was not	c. It was	d. It was slowly				

	for long time	properly	suddenly	cooled in air		
		cleaned before	cooled in brine			
		heating				
8	The process of producing a component with tough and ductile core				CO4	1
	and a hard outer	and a hard outer surface is known as				
14	a. Hardening	b. Case	c. Tempering	d. Annealing		14
.22		Hardening	N N			.22
9	The study of metallographic includes				CO5	1
10	a. alloy	b. failure	c. metal	d. all of the		
LΩ	constituents	analysis	structure	above		Ω
10	In order to obser	der microscope,	CO5	1		
	the magnificatior	n should be the ord	ler of			
	a. 2	b. 20	c. 1500	d. 100		
11	Dye penetrant m	ethod is generally	used to locate		CO6	1
	a. core defects	b. surface	c. superficial	d. temporary		
		defects	defects	defects	9 1	5 L
12	During radiography test, which region absorbs less radiation and					N 1
03	transmits more?					0.3
	a. Low and high	b. High density	c. Low density	d. None of the		10
	density regions	region	region	above		
	absorb and					
	transmit same					
	amount of					
	radiation					
4			4			+
Q. 2	Solve the following.					L 12
A)	Explain stress strain curves for different materials with figures.				CO2	6
₿	Explain Imperfections in crystals.				C01	0.0
51	Ŋ					LÇ
Q.3	Solve the following.					12
A)	Draw neat and complete Iron-iron carbide equilibrium diagram,				CO3	6
	showing phases, temperatures.					
B)	Describe classifications, and applications of different steels.				CO3	6

Q. 4	Solve Any Two of the following.			12
A)	Explain tensile testing method with stress-strain curve and necessary	CO2		6
	formulas.			
B)	Describe the normalizing heat treatment process with neat figure.	CO4		6
4	7	ł	4	
શ્	Describe the nitriding heat treatment process with neat figure.	CO4	2	6
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Ö	00		Ď	
Q.5	Solve Any Two of the following.		51	12
A)	Explain in brief how to prepare specimen for metallography testing.	CO5		6
B)	Describe Spark test with figures.	CO5		6
C)	Draw CCT diagram for steel and write its importance.	CO3		6
Q. 6	Solve Any Two of the following.			12
A)	Differentiate between destructive testing and non-destructive testing.	CO6 .	4	6
B)	Explain ultrasonic inspection with neat figure.	CO6	21	6
6	Explain any one Strengthening method for metals.	CO6	38	6
0	*** End ***	(10	
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